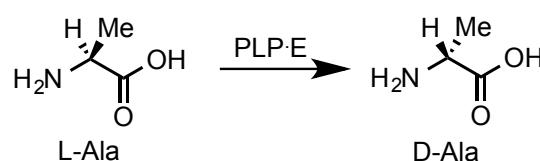


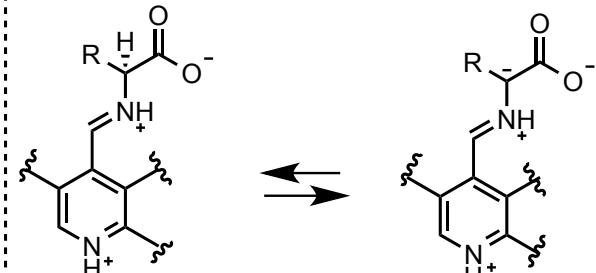
## Lecture 9 - Co-factors Part II

**PLP**

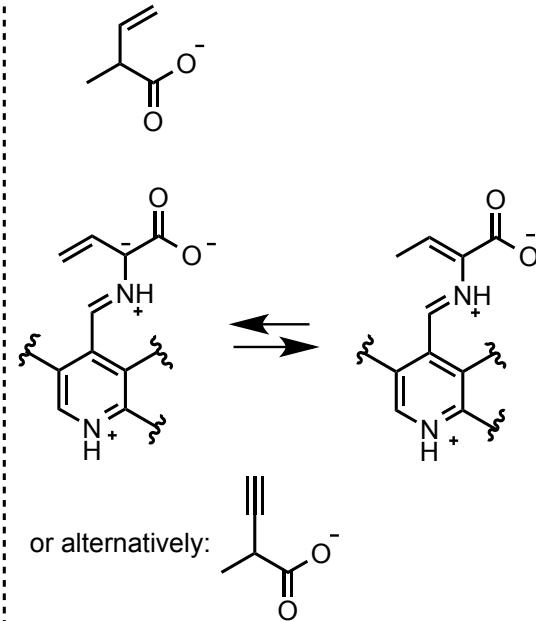
- Alanine Racemase



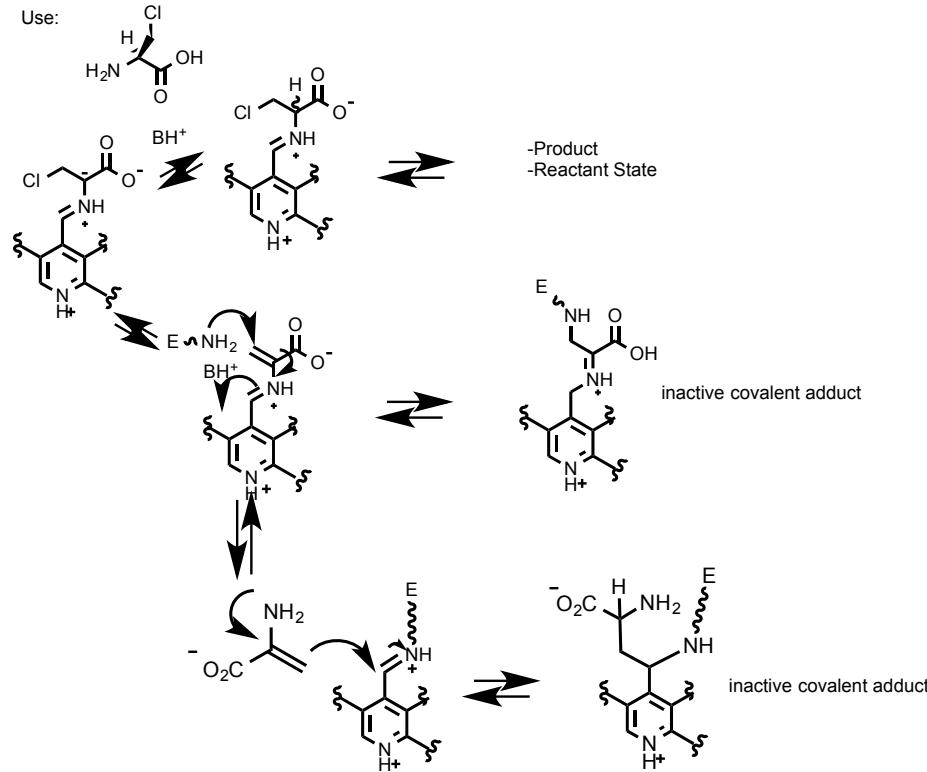
- How to inhibit the conversion of L-Ala to D-Ala?



Another covalent inhibitor:

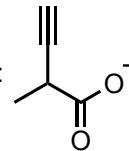


Designing a mechanism based PLP inhibitor?



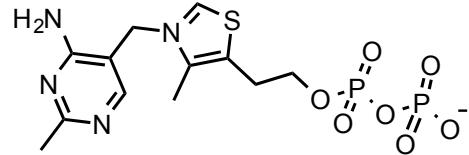
$$\frac{k_{inact}}{k_{cat}} \approx \text{Partitioning}$$

or alternatively:

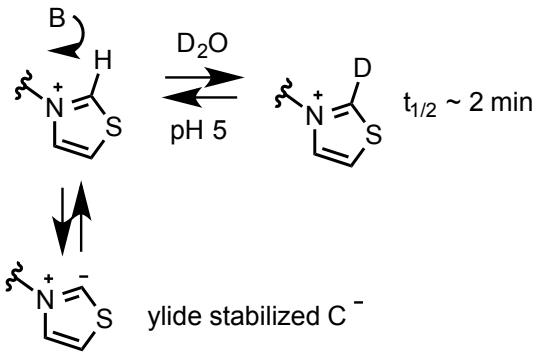


## Thiamine pyrophosphate (Vitamin B<sub>1</sub>)

Like PLP, the electron-deficient heterocyclic ring of thiamine pyrophosphate (TPP) stabilizes the formation of carbanionic species.

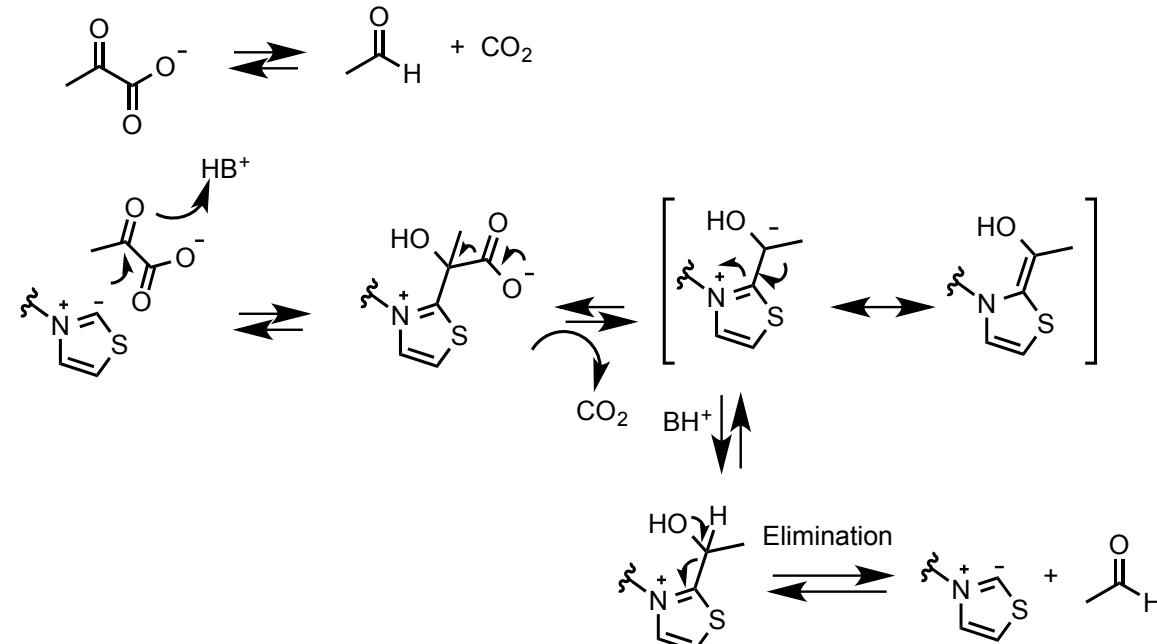


Breslow (1961) was the first to propose and provide evidence for the C-2 carbanion, showing that this proton exchanges readily in D<sub>2</sub>O

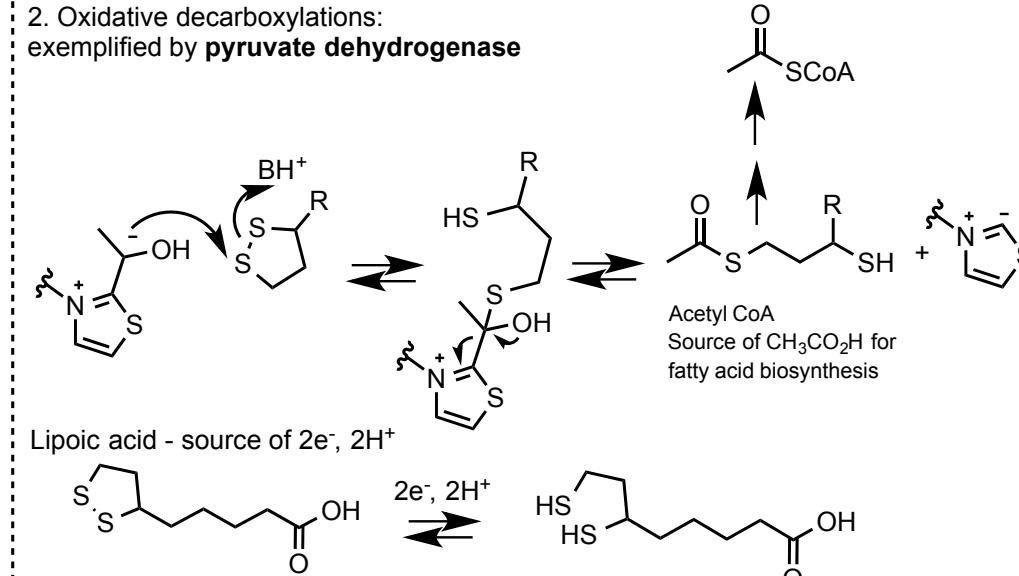


## Reactions involving thiamine pyrophosphate:

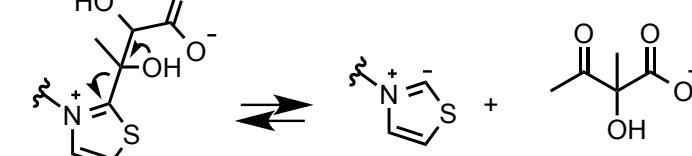
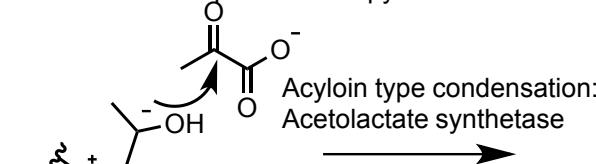
### 1. Decarboxylation of alpha-ketoacids: pyruvate decarboxylase



### 2. Oxidative decarboxylations: exemplified by pyruvate dehydrogenase



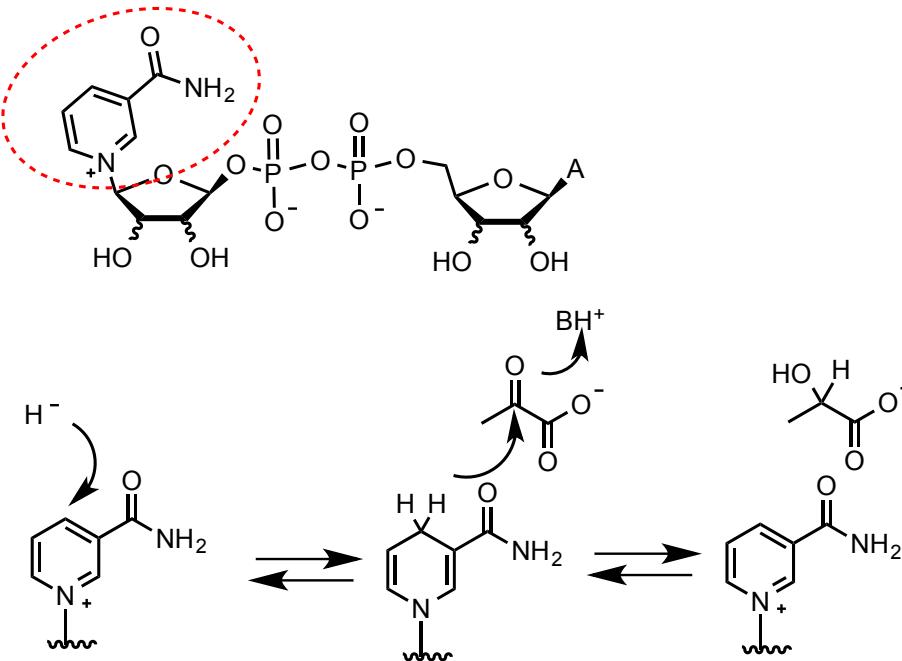
-could also intercept with more pyruvate



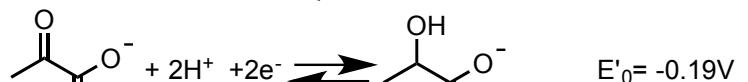
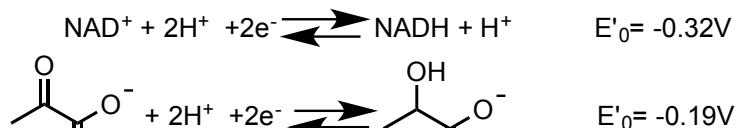
## Redox Enzymes:

- NADPH/NADH  $\rightleftharpoons$  nicotinamide dependent ( $2e^-$ ,  $H^+$ )
- FAD/FADPH  $\rightleftharpoons$  flavin dependent  $1e^- \rightarrow 1e^- (2e^-, 2 H^+)$

## Nicotinamide adenine dinucleotide (phosphate)

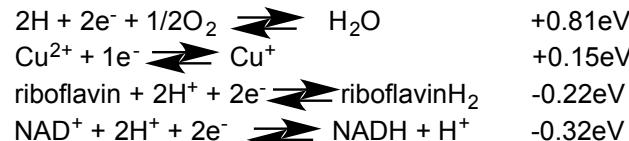
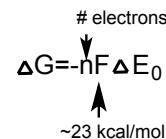


Overall Reaction:



$$E'_0 = E'_0(\text{red}) - E'_0(\text{oxid}) = -0.19\text{V} - (-0.32\text{V}) = 0.13\text{V}$$

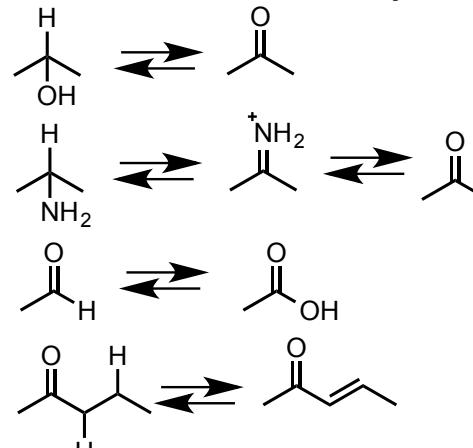
Note compounds with lower redox potential thoroughly reduce those with higher redox potentials



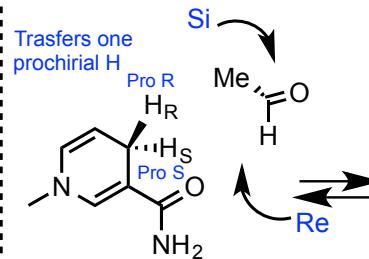
-use nicotinamide to reduce riboflavin

-molecular oxygen is reduced by riboflavin, not nicotinamide (1e<sup>-</sup> process)

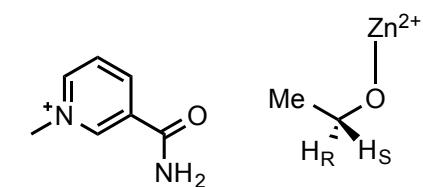
## Reactions of NAD<sup>+</sup>/NADPH<sup>+</sup> Enzymes (2e<sup>-</sup>/H<sup>+</sup>)



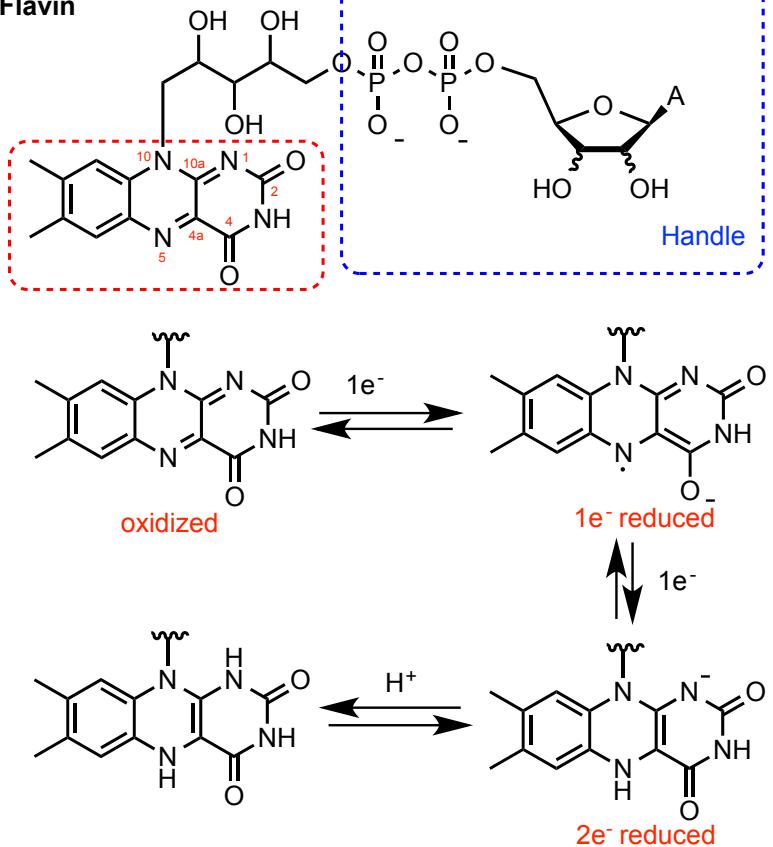
Note that these are all steriospecific



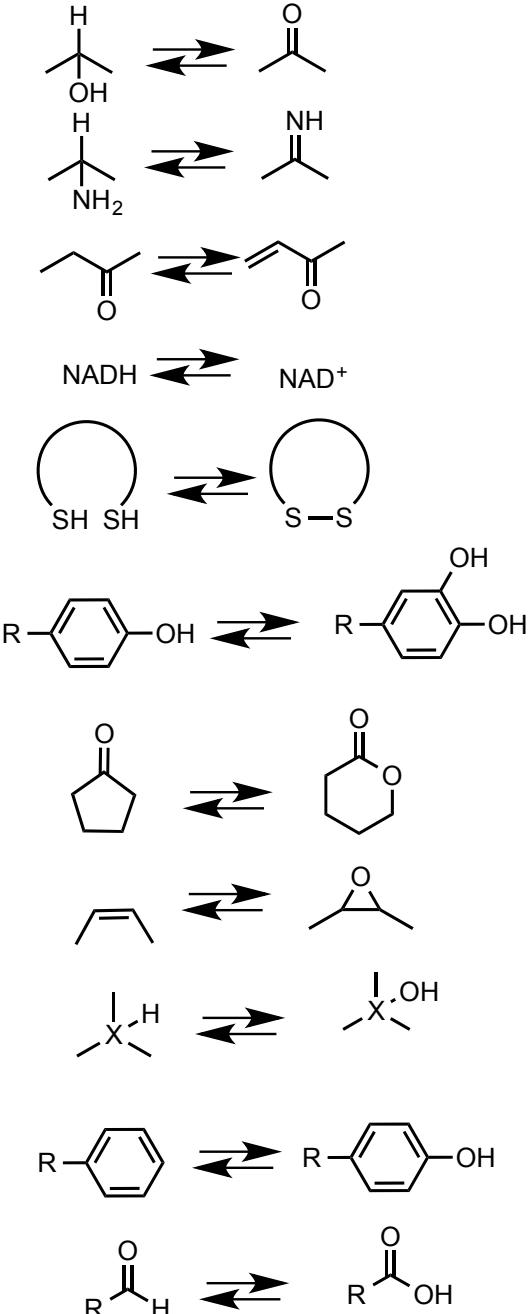
To deduce whether it is pro R or pro S replace that H by D and determine R or S configuration



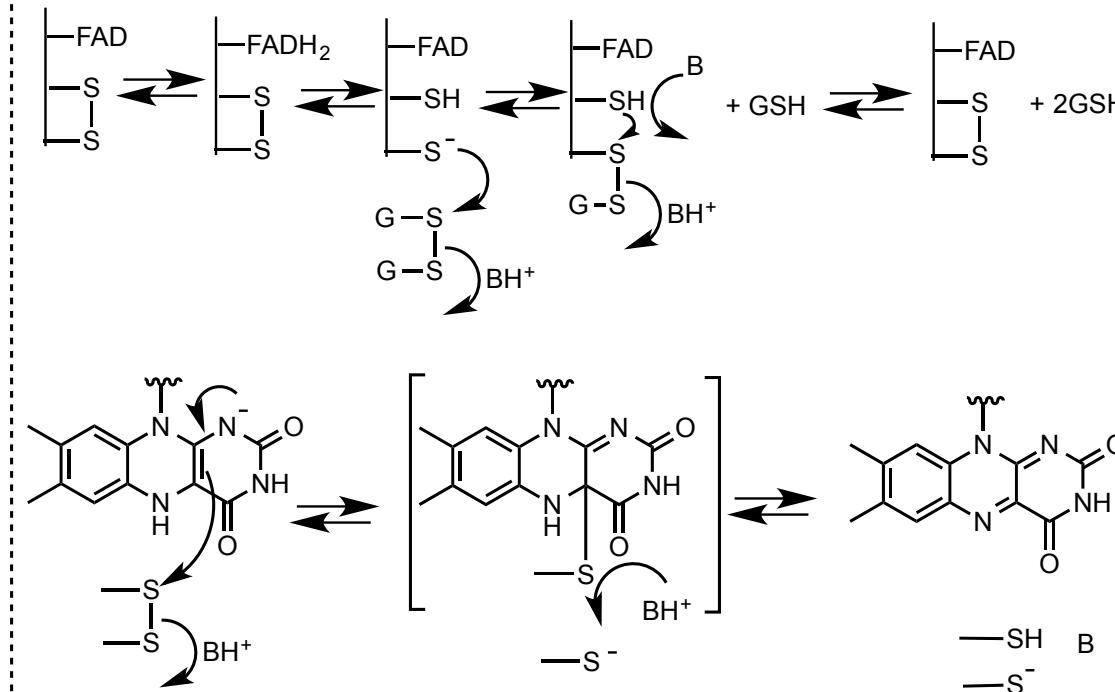
### Flavin



### Reactions of Flavin Enzymes ( $2\text{e}^-/2\text{H}^+$ )



### Reaction of glutathione (glutathione reductase)



### Molecular oxygen

